

**PCT**

WORLD INTELLECTUAL PROPERTY ORGANIZATION  
International Bureau



INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

<b>(51) International Patent Classification <sup>3</sup>:</b>  C08J 9/00; C08G 18/28	<b>A1</b>	<b>(11) International Publication Number:</b> WO 80/02427  <b>(43) International Publication Date:</b> 13 November 1980 (13.11.80)
<b>(21) International Application Number:</b> PCT/GB80/00072  <b>(22) International Filing Date:</b> 28 April 1980 (28.04.80)  <b>(31) Priority Application Number:</b> 7914847  <b>(32) Priority Date:</b> 28 April 1979 (28.04.79)  <b>(33) Priority Country:</b> GB  <b>(71) Applicant:</b> BP CHEMICALS LIMITED [GB/GB]; Britannic House, Moor Lane, London EC2Y 9BU (GB).  <b>(72) Inventor:</b> TENHAGEN, Rudolf, Johann; BP Chemicals (Suisse) S.A., 243, route des Fayards, CH-1290 Versoix/GE (CH).		<b>(74) Agent:</b> RYAN, Edward, Terrence; BP Trading Limited, Patents & Licensing Division, Chertsey Road, Sunbury-on-Thames, Middlesex TW16 7LN (GB).  <b>(81) Designated States:</b> AU, BR, JP.  <b>Published</b> <i>With international search report</i>
<b>(54) Title:</b> PROCESS FOR MAKING POLYURETHANE FOAMS  <b>(57) Abstract</b>  Cold-cured moulded polyurethane flexible foams having improved vibration damping properties are made by foaming in the presence of (1) a high molecular weight predominantly linear siloxane polyoxyalkylene block copolymer and (2) a solvent which is compatible with the block copolymer and which has a viscosity of 3 to 100 cps and a surface tension between 15 and 60 dynes/cm.		

**FOR THE PURPOSES OF INFORMATION ONLY**

**Codes used to identify States party to the PCT on the front pages of pamphlets publishing international applications under the PCT.**

<b>AT</b>	Austria	<b>LI</b>	Liechtenstein
<b>AU</b>	Australia	<b>LU</b>	Luxembourg
<b>BR</b>	Brazil	<b>MC</b>	Monaco
<b>CF</b>	Central African Republic	<b>MG</b>	Madagascar
<b>CG</b>	Congo	<b>MW</b>	Malawi
<b>CH</b>	Switzerland	<b>NL</b>	Netherlands
<b>CM</b>	Cameroon	<b>NO</b>	Norway
<b>DE</b>	Germany, Federal Republic of	<b>RO</b>	Romania
<b>DK</b>	Denmark	<b>SE</b>	Sweden
<b>FR</b>	France	<b>SN</b>	Senegal
<b>GA</b>	Gabon	<b>SU</b>	Soviet Union
<b>GB</b>	United Kingdom	<b>TD</b>	Chad
<b>HU</b>	Hungary	<b>TG</b>	Togo
<b>JP</b>	Japan	<b>US</b>	United States of America
<b>KP</b>	Democratic People's Republic of Korea		

PROCESS FOR MAKING POLYURETHANE FOAMS

The present invention relates to a process for the production of flexible polyurethane moulded foams. For comfortable transportation of seated passengers in cars, buses, trains, aeroplanes etc. a seat construction made out of springs, rubberised hair, converted conventional foam, moulded conventional or high resilient polyurethane foams, foam composite etc. is used to provide comfort in seats and backs by supporting the passenger's body, at the same time damping vibrations of different amplitudes and frequencies which exist on the seat base, caused by the movement of the car, train, bus, aeroplane etc. Depending on the type and construction of the transporter different vibrations exist which have to be controlled for comfort reasons.

It is known that flexible polyurethane foams as such have basically a good vibration damping characteristic, which is better than compositions made out of metal springs with different topper pads.

Therefore an increasing number of seats for cars, trains, aeroplanes, buses etc. are made out of urethane foams, either cut and shaped from slab-stock buns or directly moulded using the hot moulding or one shot cold cure technique. These moulded parts are used alone or on flat springs, or other flexible supports such as rubber strings. With this relatively new technique it was possible to improve the seating comfort in movable vehicles obtaining at the same time savings on total seat costs, through easier and faster assembling.

A problem which is sometimes found with the known polyurethane foams used as seats in cars, buses and other means of transport is that if a load, for example a person, supported by the foam is subjected



to vibrations at certain frequencies the amplitude of the movement by the load in response to the applied vibration increases due to reinforcement by successive vibrations until undesirably high levels are reached.

5 With the known polyurethane foam production technology it is only possible to produce very open celled or closed celled conventional or high resilient urethane foams which means that an adjustment on vibration damping properties is not possible in one step. A combination of both types, open and closed celled foam, conventional or  
10 high resilient foam is partly used to vary the vibration damping properties.

It is highly desirable to be able to provide a method of damping these undesirable vibrations. The known foaming processes produce foams with very open cells or foams with completely closed cells.  
15 Attempts have been made to overcome some of the problems with mixtures of different types of foam. It would be highly desirable however to produce a foam by a single foaming step which had improved vibration damping or control properties and which would facilitate the preparation of a range of foams with different vibration damping properties.

20 According to the present invention a process for the production of a cold-cured moulded polyurethane flexible foam by reaction of a polyol with a polyisocyanate and foaming in the presence of a blowing agent followed by crushing of the moulded foam to open the foam cells is characterised in that foaming takes place in the presence of a mixture  
25 of (1) a high molecular weight predominantly linear siloxane polyoxyalkylene block copolymer of formula  $(AB)_n$  wherein n is an integer, A is a siloxane block and B is a polyoxyalkylene block and (2) a solvent which is compatible with the high molecular weight predominantly linear siloxane polyoxyalkylene block copolymer and which has a viscosity of 3 to 1000  
30 cps, and a surface tension between 15 and 60 dynes/cm.

Processes for the production of cold cured moulded polyurethane foams in which a polyol and an isocyanate are reacted together in closed moulds in the presence of a blowing agent are well known to those skilled in the art and are to be distinguished from the processes  
35 using open conveyors by which slabstock buns are produced. The process



of the present invention may be applied to any of the conventional processes for the production of cold cured flexible moulded polyurethane foam. The pre-polymer technique can be used in which the polyol is reacted with an excess of polyisocyanate to give a pre-polymer which is foamed in a second step by the action of a blowing agent, usually water. Alternatively the one shot technique may be used in which reaction of the polyol and isocyanate and the foaming reaction take place in a single step.

#### The Polyol

In producing cellular urethane polymers in accordance with the teachings of this invention, the reaction mixture or foam formulation contains an active hydrogen-containing organic compound having an average of at least two and usually not more than five active hydrogen atoms present as hydroxyl groups. Such organic polyol reactants include compounds consisting of carbon, hydrogen and oxygen as well as compounds which contain these elements in combination with phosphorus, halogen and/or nitrogen. Suitable classes of organic polyol reactants for use in the method of this invention are polyether polyols, polyester polyols, polylactone polyols, nitrogen-containing polyols, phosphorus-containing polyols, phenolic-based polyols, and polymer/polyols produced by polymerising an ethylenically unsaturated monomer in one of the aforesaid polyols in the presence of a free radical initiator, or reacting isocyanates with primary and/or secondary amino groups containing polyamines and/or hydrazines in presence of above mentioned polyols, as described in the German Offenlegungsschrift No. 25 19 004 (4.11.76).

It is well known to the polyurethane art that the particular polyol reactant or combination of polyols employed depends upon the end-use of the polyurethane product.

For this purpose the polyol is usually characterised by its hydroxyl number which is determined by and defined as the number of milligrams potassium hydroxide required for the complete neutralisation of the hydrolysis product prepared from 1g of polyol or mixture of polyols. The hydroxyl number is also defined by the following equation which reflects its relationship with the functionality and molecular weight



of the polyol.

$$\text{OH} = \frac{56.1 \times 1000 \times f}{\text{M.W.}}$$

5 wherein

OH = hydroxyl number of polyol

f = average functionality, that is, average number of hydroxyl groups per molecule of polyol

M.W. = average molecular weight of the polyol

10

The above described polyols are used with advantage in the manufacture of high resilience foam manufacture.

15

Such foams usually have a resiliency of from about 55 to about 70 percent, as measured by standard test procedure ASTM D-3574-77. In accordance with a preferred embodiment of this aspect of the present invention, high-resilience foam formulations are used wherein at least 40 weight percent of the total polyol content is constituted of a polyether triol having the following additional characteristics:

20

(a) an average primary hydroxyl content of at least 40 mole percent (or no more than 60 mole percent of the less reactive secondary hydroxyl groups); and (b) an average molecular weight of from about 2000 to about 8000. Preferably, such polyether triols for use as

25

components of high-resilience formulations contain from about 60 to about 90 mole percent of primary hydroxyl groups and have an average molecular weight of from about 4000 to about 7000. Consistent with their trifunctionality and the aforesaid respective ranges of molecular weights, such polyether triols have hydroxyl numbers from 84 to 21, preferably from 42 to 24. These highly reactive polyether triols are provided by oxyalkylation of one of the aforesaid trihydric starters such as glycerol, with propylene oxide and ethylene oxide.

30

Usually, the total ethylene oxide content of the polyether triols is between about 7 and about 20 weight percent, expressed on the basis of total alkylene oxide fed during the oxyalkylation reaction. The high primary hydroxyl content is introduced by capping of the polyoxyalkylene chains with at least a portion of the total ethylene

35

oxide feed.



In providing high resilience foams, the polyether triols may be used as essentially the sole type of polyol in the formulation or they may be employed in combination with other polyols to control the degree of softness or firmness of the foam to vary the load bearing properties.

#### The Isocyanate

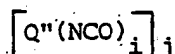
The polyisocyanate components employed in this invention for mixing with active hydrogen compounds preferably are those having the general formula:



wherein  $i$  is an integer of two or more and  $Q$  is an organic radical having the valence of  $i$ .  $Q$  can be a substituted or unsubstituted hydrocarbon group (e.g. an alkylene or an arylene group).  $Q$  can be a group having the formula  $Q' - Z - Q'$  where  $Q'$  is an alkylene or arylene group and  $Z$  is  $-O-$ ,  $-O-Q'-$ ,  $-CO-$ ,  $-S-$ ,  $-S-Q'-$ ,  $S-$ , or  $-SO_2-$ . Examples of such compounds include hexamethylene diisocyanate, 1,8-diisocyanato-p-menthane, xylene diisocyanate,  $(OCNCH_2CH_2OCH_2)_2O$ , 1-methyl-2,4-diisocyanatocyclohexane, phenylene diisocyanates, tolylene diisocyanates, chlorophenylene diisocyanates, diphenylmethane-4,4'-diisocyanate, naphthalene-1,5-diisocyanate, triphenylmethane-4,4',4''-triisocyanate, and isopropylbenzene- $\alpha$ -4-diisocyanate.

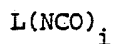
$Q$  can also represent a polyurethane radical having a valence of  $i$  in which case  $Q(NCO)_i$  is a composition conventionally known as a pre-polymer. Such pre-polymers are formed by reacting a stoichiometric excess of a polyisocyanate as set forth hereinbefore and hereinafter with an active hydrogen-containing component as set forth hereinafter, especially the polyhydroxyl containing materials or polyols.

Further included among the isocyanates useful in this invention are dimers and trimers of isocyanates and diisocyanates and polymeric diisocyanates such as those having the general formula:



in which  $i$  and  $j$  are each integers of two or more, and  $Q''$  is a poly-functional organic radical, and/or, as additional components in the

mixtures, compounds of the general formula:



in which  $i$  is one or more and  $L$  is a monofunctional or polyfunctional atom or radical. Examples of this type include ethylphosphonic diisocyanate,  $C_2H_5P(O)-N-(NCO)_2$ ; isocyanates derived from sulfonamides ( $QS O_2NCO$ ), cyanic acid, and thiocyanic acid.

More specifically, the polyisocyanate component employed in this invention also include the following specific compounds as well as mixtures of two or more of them; 2,4-tolylene diisocyanate, 2,6-tolylene diisocyanate, crude tolylene diisocyanate, bis(4-isocyanatophenyl) methane, polyphenylmethylenepolyisocyanates that are produced by phosgenation of aniline formaldehyde condensation products, dianisidine diisocyanate, toluidine diisocyanate, xylylene diisocyanate, bis(2-isocyanatoethyl) fumarate, bis(2-isocyanatoethyl) carbonate, 1,6-hexamethylene-diisocyanate, 1,4-tetramethylene-diisocyanate, 1,10-decamethylene-diisocyanate, cumene-2,4-diisocyanate, 4-methoxy-1,3-phenylene-diisocyanate, 4-chloro-1,3-phenylenediisocyanate, 4-bromo-1,3-phenylenediisocyanate, 2-ethoxy-1,3-phenylenediisocyanate, 2,4'-diisocyanatodiphenylether, 5,6-dimethyl-1,3-phenylenediisocyanate, 2,4-dimethyl-1,3-phenylenediisocyanate, 4,4'-diisocyanatodiphenylether, bis 5,6-(2-isocyanatoethyl)(bicyclo 2.2.1)hept-2-ene, benzidenediisocyanate, 4,6-dimethyl-1,3-phenylene 9.10-anthracenediisocyanate, 4,4'-diisocyanatodibenzyl, 3,3-dimethyl-4,4'-diisocyanatodiphenylmethane, 2,6-dimethyl-4,4'-diisocyanatodiphenyl, 2,4-diisocyanatostilbene, 3,3'-dimethyl-4,4'-diisocyanatodiphenyl, 3,3'-dimenthoxy-4,4'-diisocyanatodiphenyl, 1,4-anthracene diisocyanate, 2,5-fluorenediisocyanate, 1,8-naphthalene diisocyanate, 2,6-diisocyanatobenzfuran, 2,4,6-toluene diisocyanate, and many other organic polyisocyanates that are known in the art, such as those that are disclosed in an article by Siefken, Ann. 562, 75 (1949).

#### Cross-linking Agents

Further modifications are possible by adding cross-linking agents from 0.2 to 20 pphp preferable 0.5 to 8 pphp to vary hardness density ratios and/or improve specific physical properties.





Cross-linking agents are two and or higher functional primary and/or secondary amines and/or polyalcohols having molecular weights from 60 to 2000 molecular preferable 80 to 400.

The following cross-linking agents can be used: aliphatic, alicyclic, cycloaliphatic and aromatic amines, such as ethylenediamine, 1.2- and 1.3-propylene diamine, tetramethylenediamine, hexamethylenediamine, dodecamethylenediamine, trimethyldiaminohexane, N,N'-dimethylethylenediamine, 2,2'-bisaminopropyl melamine also higher homologs of ethylenediamine like diethylenetriamine, triethylenetetramine and tetraethylenepentamine, homologs of propylenediamines like dipropylenetriamines, piperazine, N,N'-bisaminoethylpiperazine, triazine, 4-aminobenzylamine, 4-aminophenyl ethylamine, 1-amino-3.3.5-trimethyl-5-aminomethyl-cyclohexane, 4,4'-diaminodicyclohexyl-methane and -propane, 1.4-diaminecyclohexane, phenylenediamine, naphthylenediamine, condensates from aniline, formaldehyde, toluenylenediamines, bis-aminomethylbenzole and other single or double monoalkylated aromatic amines.

Hydroxyl groups-containing cross-linking agents are: ethanediol, propanediol-1.2, butanediol 1.2, -1.3, -1.4, hexandiol, diethylene-glycol, triethyleneglycol, tetraethyleneglycol, dipropyleneglycol, tripropyleneglycol hydroquinone.

#### Blowing Agents

Water and/or readily volatile organic substances are used as blowing agents in the process according to the invention.

Generally, the blowing agent is employed in an amount from about 1 to about 15 parts by weight per 100 parts by weight of total polyol reactant, the particular blowing agent and amount thereof depending upon the type of foam product desired. Flexible foam formulations usually contain no more than about 6 pphp of water. The selection and amount of blowing agent in any particular foam formulation is well within the skill of the cellular polyurethane art. Suitable organic blowing agents are e.g. acetone, ethyl acetate, halogenated alkanes, such as methylene chloride, chloroform, ethylidene chloride, vinylidene chloride, monofluorotrichloromethane, chlorofluoromethane or dichlorodifluoromethane, butane, hexane, heptane or diethyl ether. A blowing effect can also be



obtained by adding compounds which decompose at temperatures above room temperature to liberate gases, e.g. azo compounds such as azoisobutyric acid nitrile which liberate nitrogen. Further examples of blowing agents and details of the use of blowing agents may be found in Kunststoff-Handbuch, Volume VII, published by Vieweg and Hoechtlen, Carl-Hanser-Verlag, Munich 1966 e.g. on pages 108 and 109, 453 and 507 to 510.

#### Catalysts

Catalysts are also frequently used in the process according to the invention. The catalysts used are known per se, e.g. tertiary amines such as triethylamine, tributylamine, N-methyl-morpholine, N-ethyl-morpholine-N-Cocomorpholine, N,N,N',N'-tetramethyl-ethylene-diamine, 1,4-diaza-bicyclo-(2,2,2)-octane, N-methyl-N'-dimethyl-amino-ethyl-piperazine, N,N-dimethyl benzylamine, bis-(N,N-diethylamino-ethyl)-adipate, N,N-diethylbenzylamine, pentamethyl diethylenetriamine, N,N-dimethylcyclohexylamine, N,N,N',N'-dimethyl-phenylethylamine, 1,2-dimethyl imidazole and 2-methyl-imidazole, triethylene diamine, bis(2-dimethylamino ethyl) ether.

The tertiary amines which contain hydrogen atoms capable of reacting with isocyanate groups may be e.g. triethanolamine, triisopropanolamine, N-methyldiethanolamine, N-ethyldiethanolamine, N,N-dimethylethanolamine or their reactions products with alkylene oxides such as propylene oxide and/or ethylene oxide.

Silaamines which contain carbon-silicon bonds may also be used as catalysts, e.g. the compounds described in German patent specification No. 1 229 290 such as 2,2,4-trimethyl-2-silamorpholine or 1,3-diethylaminomethyl-tetramethyl-disiloxane.

The catalysts used may also be bases which contain nitrogen such as tetraalkyl ammonium hydroxides or alkali metal hydroxides such as sodium hydroxide, alkali metal phenolates such as sodium phenolate or alkali metal alcoholates such as sodium methylene. Hexahydrotriazines may also be used as catalysts.

Organic metal compounds may also be used as catalysts according to the invention, especially organic tin compounds.

The organic tin compounds used are preferably tin (II) salts



of carboxylic acids such as tin (II)-acetate, tin (II) octoate, tin (II)-ethylhexoate and tin (II)-laurate and the dialkyl tin salts of carboxylic acids such as dibutyl tin diacetate, dibutyl tin dilaurate, dibutyl tin maleate or dioctyl tin diacetate.

5 Other examples of catalysts which may be used for the process according to the invention and details of their mode of action are described in Kunststoff-Handbuch, Volume VII, published by Vieweg and Hoechtlen, Carl-Hanser-Verlag, Munich 1966, e.g. on pages 96 to 102.

10 The catalysts are generally used in a quantity of between 0.001 and 10% by weight, based on the quantity of compounds which contain at least two hydrogen atoms capable of reacting with isocyanates.

Optional Conventional Organosiloxane Surfactant

15 In producing cellular polyurethanes in accordance with the method of this invention, a minor amount of an organosilicone surfactant may also be present as an additional component of the polyurethane-forming reaction mixture. When used such surfactants are usually present in amounts up to about 5 parts by weight per 100 parts by weight of total polyol reactant. Suitable classes of  
20 silicone surfactants are the polysiloxane-polyoxyalkylene block copolymers wherein the respective blocks are joined through silicon-to-carbon or silicone-to-oxygen-to-carbon bonds and the respective polyoxyalkylene blocks are bonded to different silicone atoms of the polysiloxane backbone to form a comb-like structure. Usually,  
25 the polysiloxane blocks are trialkylsiloxyl end blocked. In addition to the siloxyl units to which the pendant polyoxyalkylene chains are bonded, the polysiloxane backbone is formed of difunctional siloxyl units wherein the respective two remaining valences of silicon are satisfied by bonds to organic radicals. Illustrative of such organic  
30 radicals are the hydrocarbyl groups having from 1 to 12 carbon atoms including alkyl, aryl, aralkyl, bicycloheptyl and halogen substituted derivatives of such groups. The polyoxyalkylene blocks are usually constituted of oxyethylene units, oxypropylene units or a combination of such units, and the polyoxyalkylene chains are hydroxyl-terminated  
35 or capped with a monovalent organic group such as alkyl, aryl, aralkyl,

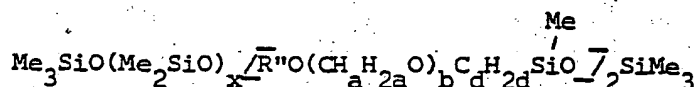


acyl, carbamyl and the like. Especially useful as stabilisers of flexible polyether-based polyurethane foams are the block copolymers described in US patent No. 3 505 377 and US Reissue Pat. 27 541.

5 The copolymers of the latter patent contain from 40 to 200 dimethyl-siloxyl units as essentially the sole type of difunctional unit, and from 15 to 60 weight percent of the oxyalkylene content of the poly-oxyalkylene blocks is constituted of oxyethylene.

The process of the present invention is preferably applied to the production of high resilience foams.

10 Because of the high reactivity of high-resilience foam formulations, the foams are generally self-stabilising and can be obtained without the use of stabilising agents. However, it is usually desirable to include a silicone surfactant as an additional component of such formulations in order to minimise the tendency of the foam  
15 to settle and to control cell uniformity. Particularly effective for this purpose are the relatively low molecular weight polyoxyalkylene-polysiloxane block copolymers described and claimed in US patent No. 3 741 917. Especially suitable as components of high-resilience formulations are the block copolymers described therein having the  
20 formula:



wherein x has an average value of from 2 to 7; b has a value from 3 to 10; z has an average value from 2 to 6; a and d each has a value from 2 to 4; and R'' is a monovalent hydrocarbon radical such  
25 as alkyl, aralkyl and aryl radicals, or an acyl group.

Also suitable as organosilicone components of high-resilience foam formulations are the relatively low molecular weight aralkyl-modified polymethylsiloxane oils described and claimed in US patent No. 3 839 384.

30 The organosilicone component is usually present in high-resilience formulations on an amount between about 0.025 and about 2 parts by weight per 100 parts by weight of total polyol.

#### Additional Ingredients

35 It is also sometimes desirable to include various additives in the reaction mixture such as colouring agents, fillers, flame retardants



and the like. Suitable colouring agents are, for example carbon black, titanium dioxide, menthol blue, chromium red and the like. Suitable fillers are fatty acids including tall oil fatty acids, or tall oil per se, which, if desired, may be halogenated for example, with chlorine or bromine, vermiculite, saw dust, synthetic plastics including vinyl polymers such as, polyvinyl chloride, polystyrene and the like. Suitable flame retardants are antimony oxide, tris (chloroethyl) phosphate, tricresyl phosphate, triphenyl phosphate and the like.

#### The Novel Additives

#### The Siloxane Block Copolymers

The high molecular weight predominantly linear siloxane polyoxyalkylene block copolymers used in the control of the vibration characteristics of the present invention are different from the polysiloxane polyoxyalkylene block copolymers conventionally used as foam stabilisers in chemically blown polyurethane foams. They differ from the conventional foam stabilisers in being predominantly linear and in having a higher molecular weight. They are in general stronger surfactants than the foam stabilisers conventionally used in chemically blown moulded foams.

The high molecular weight siloxane polyoxyalkylene block copolymers are of formula  $(AB)_n$  where  $n$  is an integer,  $A$  is a siloxane block and  $B$  is a polyoxyalkylene block. The copolymers are predominantly linear in that the chain of repeating units  $A$  and  $B$  is predominantly linear rather than predominantly branched. This does not however exclude the presence of branching in the individual repeating units  $A$  and  $B$ , particularly in the siloxane unit  $A$ .

The predominantly linear high molecular weight siloxane polyoxyalkylene block copolymers may be divided into two classes.

1. Hydrolysable block copolymers, i.e. those wherein the siloxane blocks and the polyoxyalkylene blocks are linked by silicon to oxygen to carbon linkages and/or by means of oxygen bondings.
2. Non-hydrolysable (hydrolytically stable) block copolymers, i.e. those wherein the siloxane blocks and polyalkylene blocks are linked by silicon to carbon linkages.

The preferred block copolymers are those of the hydrolysable type.



The average molecular weight of each siloxane block of the siloxane polyoxyalkylene or block copolymers is preferably in the range from about 500 to 10,000. The monovalent hydrocarbon radicals need not be identical throughout the siloxane block, but can differ from one  
5 siloxane unit to the next and can even be different within a single siloxane unit.

The predominantly linear polyoxyalkylene block copolymers can have an average molecular weight of about 20,000 to 250,000 or higher. Preferably the hydrolysable type  $(AB)_n$  polymers have an average molecular  
10 weight of at least about 65,000 and most preferably about 100,000, while the non-hydrolysable type  $(AB)_n$  polymers have an average molecular weight of at least 20,000. It is to be understood that the above molecular weight ranges apply to the high molecular  $(AB)_n$  copolymers. When the  
15 high molecular weight  $(AB)_n$  copolymers are employed in admixture with lower molecular weight materials resulting from the use of mono-ols in admixture with the diols, the average molecular weight of the total product may be lower than the values given above

The high molecular weight predominantly linear siloxane polyoxy-alkylene block copolymers are most preferably used in admixture with  
20 low molecular weight siloxane polyoxyalkylene condensation products obtained by reaction of polydialkyl siloxane with a mixture containing mono-ol in addition to the diol.

The high molecular weight predominantly linear siloxane polyoxy-alkylene block copolymers may be made by the reaction of (1) a polydialkyl  
25 siloxane which has been end-blocked with alkoxy groups with (2) a polyoxyalkylene diol.

In order to prevent excessive branching of the block copolymer chain and to give a predominantly linear product it is highly desirable to react the end-blocked polydialkyl siloxane with a mixture of diol  
30 and mono-ol. The mono-ol is preferably a mono-ether of a polyoxyalkylene ether diol. The polyoxyalkylene diol from which the mono-ether is derived may be the same polyalkylene ether diol which is reacted with the polydialkyl siloxane or may be different.

The weight ratio of diol to mono-ol is preferably in the range 1:1  
35 to 1:2.



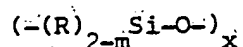
The weight ratio of total diol and mono-ol to polyalkyl siloxane may for example be in the range 7:1 to 1:1, preferably 5:1 to 2:1. The diol may, for example, have a molecular weight in the range 1500 to 7000, preferably not less than 2500, more preferably not less than 3000, and preferably not more than 6000, more preferably not more than 4000. It may be derived, for example, from ethylene oxide or propylene oxide or mixtures of these oxides.

The mono-ol may, for example, have a molecular weight in the range 500 to 2000, more preferably 800 to 1500, and may also be derived from ethylene oxide and/or propylene oxide units. In order to give the required mono-ol one of the terminal hydroxy groups of a polyalkylene ether diol may be end-blocked with an alkoxy group, e.g. methoxy or ethoxy.

The use of a mixture of diol and mono-ol gives a predominantly linear (AB)<sub>n</sub> block copolymer and serves to substitute partially active positions in the polydialkyl siloxane so preventing cross linking reactions and the formation of gel. Additionally it dilutes the high molecular weight (AB)<sub>n</sub> block copolymer with low molecular weight structures derived from the mono-ols (mono-functional polyethers). The reaction time is shortened, and the viscosity of the final product containing the (AB)<sub>n</sub> copolymer is reduced by the low molecular weight material, which facilitates the use of the product in the process of the invention.

The end-blocked polydialkyl siloxane with which the diol is reacted may be produced by methods well known to those skilled in the art. The polydialkyl siloxane may be linear or branched, and may be derived from the equilibration product between cyclic dialkyl siloxane, e.g. cyclic dimethyl siloxane and trialkoxy alkyl silane, e.g. triethoxy methyl silane, or cyclic dialkyl siloxane and SiH fluid.

The resulting equilibrate is comprised of compounds of various functionality in the range 2 to 6. The backbone of the silicone unit is described by the average formula:-

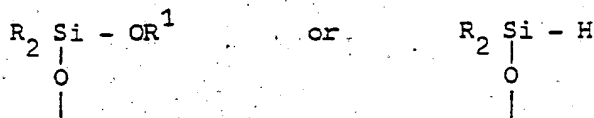


where x is an integer of at least 6, R is a monovalent hydrocarbon group and m is 0 or 1. When m is 1, the remaining valence of the Si atom is satisfied by H or by another siloxane group.

The backbone may be terminated by alkoxy groups or H.



Thus the active positions to be substituted by polyether are:



5 where  $R^1$  is a hydrocarbyl group, e.g. alkyl and aralkyl groups. Examples of suitable alkyl groups are methyl, butyl and benzyl groups.

Note that in the above formula the backbone corresponds to the case when  $m$  is 0 in the formula  $(-(R)_{2-m} \text{ Si-O-})_x$  given above.

10 It is to be understood that while said  $(AB)_n$  block copolymers used in this invention can be discrete chemical compounds they are usually mixtures of various discrete block copolymers species due at least in part to the fact that the siloxane and polyoxyalkylene reactants used to produce said  $(AB)_n$  block copolymers are themselves usually mixtures. Moreover, while a single  $(AB)_n$  block copolymer can be employed  
15 as foam cell modifier, mixtures of two or more different  $(AB)_n$  block copolymers can also be employed if desired.

The quantity of the linear block copolymer added can vary over a moderately wide range. Examples of suitable quantities are those in the range 0.05 to 5%.

## 20 The Solvent

Solvents which are used as diluents for the organic polysiloxane surfactant in this invention are compatible with the surfactant, and have a viscosity of 3 to 1000 cps preferably 10 to 500 cps. The solvent may, for example, have one to six reactive OH-group per molecule, but  
25 also neutral diluents which can be used. Solvents which cause a slight swelling effect on ready reacted polyurethane foam are preferred due to easier crushing if used in the production process of moulded polyurethane foams. The surface tension of the diluent is between 15 and 60 dynes/cm preferably between 30 and 40 dynes/cm. The viscosity is measured at 22°C.

30 Mixtures of surfactants and solvents such as alcohols, poly-alcohols polyesters, glycols, polyglycolethers, polyglycolesters, ethoxylated- and butoxylated- alcohols and glycols, aliphatic- and aromatic solvents in a ratio of from 1:99 to 99:1 by weight and every ratio in between allows the regulation of foam cell structure and openness of the cell  
35 membranes especially. Preferred weight ratios of the high molecular weight linear block copolymers to solvent are 2:100 to 10:100.





### The Foaming Step

The foaming step is carried out in a known manner in a closed mould. The foaming step can conveniently be carried out by feeding measured quantities of (1) the polyol, (2) isocyanate and (3) a mixture of the high molecular weight predominately linear siloxane polyalkylene block copolymer and solvent separately to a continuous mixer and is then poured into the mould. The cross-linking agents, blowing agents, catalysts, optional conventional organosiloxane surfactant and additional ingredients referred to above are conveniently fed to the moulding process in the polyol.

### The Crushing Step

In the production of foam by the cold cured process it is necessary for processing reasons to carry out the foaming process so as to give a proportion of closed cells in the foam. The "cold cure" process in contrast to the hot moulded process does not use a highly heated mould but the reaction is exothermic and the foam is warm when removed from the mould. To avoid distortion of the foam it is necessary to crush the foam to open the cells. The crushing step is well known to those skilled in the art and the optimum crushing conditions for use in the present invention can be easily determined by simple tests.

The invention will now be illustrated by reference to the following Examples. In these Examples measurements were made of surface porosity and core breathability of the foam. Surface breathability was measured by the Gurley test using the Gurley-Hill S-P-S Tester. This equipment was originally developed to test the printing quality of paper. It measured smoothness, porosity and softness of the paper. As no test method exists to measure the surface porosity of moulded polyurethane foams, this equipment was used to evaluate the surface porosity. The Gurley-Hill S-P-S Tester measures a given air-leakage through the foam. Air, at a low uniform pressure, is supplied by an inverted cylinder, carefully standardised as to volume and weight, floating freely in an outer cylinder partly filled with oil. The air is forced downward, through an open tube, to the upper orifice plate against which the sample is clamped. Each 50 cc. is marked by rings on the inner cylinder from 0 to 300 cc.



The test consists of timing the rings as they drop past the upper edge of the outer cylinder, using a stop watch or other timing device. Uniform clamping is provided by compressing the upper orifice plate 0.5 cm into the foam. The result is given as

$$5 \quad \frac{x}{\text{sec/300 cm}^3}$$

The core breathability corresponds to the Air Flow as measured in accordance with ASTM D 3574-77 and was determined using the well known Nopco apparatus.

10 All tests were carried out under ambient conditions. Approximately 50 to 60% relative humidity and 20 to 25°C.

The test specimen was either a square foam sample of a size of 40 x 40 x 10 cm or a complete seat.

#### Determination of Amplitude Enlargement Factor

15 This test was carried out with a plate which is connected to an eccentric. Different amplitudes and speeds (frequencies) can be simulated with the plate.

Amplitudes used in these tests are between 6 and 12 mm, at frequencies of 1 to 15 Hz.

20 On top of the plate, the foam sample (seat) to be tested is placed with a "Dummy" or weight which represents a seated person.

The enlargement factor can be calculated as follows:

$$Q_x = \frac{x_r \text{ (mm)}}{x_i \text{ (mm)}}$$

$Q_x$  = amplification quotient

25  $x_r$  = resonance amplitude

$x_i$  = initial amplitude

The results are given as a resonance maximum  $Q_x$  at a certain frequency.

#### Preparation of High Molecular Weight Surfactants

##### 30 Example A

This example illustrates the production of a high molecular weight hydrolysable siloxanepolyoxyalkylene block copolymer derived from ethoxy blocked polydimethyl siloxane (EBDMS) and a blend of mono and dihydroxy polyethers.

35 In a two-litre flask equipped with fractionating column, thermometer, stirrer and a nitrogen inlet tube for sparging there was placed 133.2 g of dihydroxy polyether having an average molecular weight of 3100 and being



composed from ethylene oxide (EO) and propylene oxide (PO) units in wt. ratio 50:50, 189 g of monohydroxy polyether of an average mol. wt. 1097 and EO/PO = 50/50, 58 g of EBDMS having an average functionality 3 and containing 8% by wt. of terminal ethoxy groups, 0.33 g dry potassium acetate and 380 g of fresh toluene.

The mixture was heated and 0.43 ml of the trifluoroacetic acid was added when the temperature reached 60°C. The temperature of the reaction mixture was set at 114°C. The azeotropic mixture of boiling point 75.5°C appears in a short time in the still head. Having fixed the dropping rate of condensate the reflux ratio was set as 1/40 and the collection of distillate was started. The temperature rose to 108.5°C and then 118 g of EBDMS in 114 g of toluene was admitted. The collection of newly formed azeotrope was pursued until 108.5°C. At that moment 6 ml of dicyclohexylamine was added and the mixture subjected to the stripping of toluene. The removal of toluene was accomplished by means of heating and rigorous nitrogen sparge.

#### Example B

A hydrolysable block polymer was prepared in the equipment described in the former example.

58 g of the silicone fluid (EBDMS) as in the former example was loaded together with 63 g of monohydroxy polyether of an av. M.W. 1097 and 50 g of polymer polyol (diol), containing 20% of styrene/acrylonitrile copolymer and having an av. M.W. 3500. 171 g of fresh toluene and 0.24 g of  $\text{CH}_3\text{COOK}$  were added as well. The mixture was heated and the catalyst:  $\text{F}_3\text{CCOOH}$  admitted at 60°C. The flask temperature was set at 114°C. The azeotrope was collected until the temperature in the still head reached 108°C. 2 g of dicyclohexylamine was then added in order to neutralise the acid. The removal of toluene was accomplished by means of heating and rigorous nitrogen sparge.

#### Example 1

A polyol mixture is prepared by mixing about 100 parts of polyether polyol having a molecular weight of 5000 and a hydroxyl number of about 34, with about 2.8 parts of water, further with about 0.5 parts of catalyst such as triethylene diamine and about 1 part of a siloxane surfactant as a stabiliser.



This blend was mixed, using a three stream moulding machine, together with about 35 parts of an isocyanate blend made from 80% toluene diisocyanate and 20% m-phenylene diisocyanate and 2 parts of a mixture made from 96% of butoxytriethylene glycol and 4% siloxane polyether block copolymer (AB)<sub>n</sub> type and about 150,000 to 200,000 M.W.

The butoxytriethylene glycol had a viscosity of 9 to 10 cSt at 25°C and a surface tension of about 36 to 38 dynes/cm at 22°C. The blend of solvent and surfactant had a viscosity of 16.6 cSt and 25°C and a surface tension of 27.5 dynes/cm at 21.7°C.

This complete mixture was poured into a closed mould made from cast aluminium and the moulded part demoulded after approximately 10 minutes.

After crushing, the resulting moulded foam part had a density of about 45 kg/m<sup>3</sup>.

TABLE 1

Foam Characteristics	Test A	Example 1
VC-agent	0 php	2.0 php
Hardness, N (IFD, ASTM)	250	290
VC-factor, Q <sub>x</sub>	3.8	2.7
Breathability Surface (Gurley) sec.	2	4
Core (NOPCO) SCFM	3.5	0.7
Cell size, mm	0.2-2.0	1.5-3.5
Cell volume, mm <sup>3</sup>	0.1-5	1.5-22
No. of breaks in cell wall	8-14	6-12
Size of breaks in cell wall, mm	0.1-1.3	0.1-0.8
Internal residue cell surface, %	50-60	70-80

Test A

In a comparative test not according to the invention Example 1 was repeated but without the mixture of butoxytriethylene glycol and siloxane polyether block copolymer.



Example 2

A commercially available urethane pre-polymer system consisting of components A and B was mixed together in a three stream moulding machine together in a ratio 1:1 plus 2% of a mixture made from 96% of butoxytriethylene glycol and 4% siloxane polyether block copolymer, (AB)<sub>n</sub> type and about 150,000 to 200,000 M.W. The ready made mixture was poured into a closed mould as mentioned in Example 1.

The resulting moulded foam part had a density of about 45 kg/m<sup>3</sup> after crushing.

TABLE II

Foam Characteristics	Test B	Example 2
VC-agent	0 php	2.0 php
Hardness, N: (IFD, ASTM)	410	440
VC-factor, Qx	3.8	2.7
Breathability: Surface (Gurley), sec:	4	5
Core (NOPCO) SCFM:	2.0	0.8
Cell size, mm:	0.2-2.0	0.6-3.5
Cell volume, mm <sup>3</sup>	0.1-5	0.5-22
No. of breaks in cell wall:	8-14	6-10
Size of breaks in cell wall, mm:	0.1-0.8	0.1-0.6
Internal residue cell surface, %:	40-50	70-80

Test B

In a comparative test not according to the invention an experiment was carried out as in Example 2 but without the mixture of butoxytriethylene glycol and siloxane polyether block copolymer.

Example 3

A polyurethane foam was prepared in the manner described in Example 1. The foam employed was produced with the following ingredients:



	<u>Ingredients</u>	<u>Parts by Weight</u>	<u>Stream</u>
	Polyol X	100 )	
	Water	2.8 )	
5	Silicone surfactant	1.0 )	I
	Triethylene diamine	0.2 )	
	Bis(2 dimethylaminoethyl)ether	0.1 )	
	Dibutyltindilaurate	0.02 )	
	Vibration Control Agent	)	
	(96% Butoxytriethylene glycol)	2.5 )	
10	(4% Siloxane polyether block copolymer, (AB) <sub>n</sub> type about 150,000 to 200,000 M.W.	)	II
	Toluene diisocyanate (80/20)	29.9 )	
	m-Phenylenediisocyanate	7.5 )	III

15 Polyol X is an acrylonitrile/styrene grafted polyol with a molecular weight of about 5000 and a hydroxyl number of about 32.

The resulting moulded foam parts had a density of about 45 kg/m<sup>3</sup> after crushing

#### Test C

20 This is a comparative example not according to the invention.  
An experiment was carried out as in Example 3 but without using the vibration control agent.

TABLE III

	<u>Foam Characteristics</u>	<u>Test C</u>	<u>Example 3</u>			
25	VC-agent	<u>0 php</u>	<u>1.5 php</u>	<u>2.0 php</u>	<u>2.5 php</u>	
	Hardness, N: (IFD, ASTM)	450	450	460	480	
	VC-factor, Qx	3.9	3.1	3.0	2.2	
	Breathability:					
30	Surface (Gurley), sec:	1	2	3	6	
	Core (NOPCC), SCFM:	4	0.6	0.5	0.2	
	Cell size, mm :	0.2-2.0	0.6-3.0	0.8-3.0	1-3.5	
	Cell volume, mm <sup>3</sup>	0.1-5	0.2-15	2.0-18	4-23	
	No. of breaks in cell wall:	8-14	8-12	8-10	6-10	
35	Size of breaks in cell wall, mm:	0.1-1.5	0.1-1.0	0.1-0.17	0.1-0.6	
	Internal residue cell surface, %	20-40	40-70	60-80	70-90	



Example 4

A series of polyurethane foams were prepared in the manner described in Example 1. The foams employed were produced with the following ingredients varying the type of solvent in the Vibration Control Agent.

5	<u>Ingredients</u>	<u>Parts by Weight</u>	<u>Stream</u>
	Polyol X	100 )	
	Water	2.6 )	
	Silicone Surfactant	0.8 )	I
	Triethylene diamine	0.33 )	
10	Bis(2-dimethylaminoethyl)ether	0.15 )	
	Vibration Control Agent	2.0 )	
	(97% of varying Solvents (see Table IV)	)	II
	(3% of Siloxane block copolymers, (AB) type and about 150,000 to 200,000 M.W.)	)	
15	Toluene diisocyanate 80-20	28.16 )	
	Crude Phenylenediisocyanate	7.04 )	III

Polyol X is an acrylonitrile/styrene grafted polyol with a molecular weight of about 5000 and a hydroxyl number of about 32.

The resulting moulded foam parts had a density of about 40 kg/m<sup>3</sup> after crushing.

TABLE IV

Type of Solvent	Surface Tension	Hardness ASTM 50%,N	VC-factor Qx	Breathability Core NOPCO/SCFM	Cell structure $\phi$ mm
Polypropylene glycol MW 1025	-	300	3.4	1.8	0.2 to 2.0
Nonionic liquid Alcohol (with 3 Moles ethylene oxide)	-	300	2.7	0.8	0.2 to 2.5
Nonionic liquid Alcohol (with 9 Moles ethylene oxide)	29	320	2.8	0.8	0.2 to 2.5
Acetone	23.7/20°C	370	2.2	0.6	0.4 to 3.0
Methylethylketone	23.5/20°C	350	3.0	0.8	0.3 to 3.0
Xylene	-	340	2.8	0.8	0.3 to 3.0
Methylenchloride	-	330	1.6	0.8	0.4 to 3.5
Fluorocarbon-11	-	300	2.0	1.1	0.5 to 3.5
Ethylalcohol	22.7/20°C	230	2.2	0.9	0.3 to 3.0
Butylalcohol	22.4/20°C	250	2.2	0.8	0.3 to 3.0
Methoxyethylacetate	-	350	2.0	0.8	0.2 to 2.5
Alkylsulphonic acid ester of phenol/cresol	-	310	3.0	1.2	0.2 to 2.5
Dimethylformamide	-	340	2.1	0.8	0.3 to 3.0
Polyether polyol 5000 MW	-	340	2.2	1.0	0.3 to 3.0
Polyether polyol 3500 MW	-	360	1.8	0.6	0.3 to 3.0



Example 5

A cured polyurethane foam was prepared using a commercial low pressure moulding machine for making HR (high resilience) polyurethane foam at a throughput of about 14 kg/min. The machine used was sold under the Trade Name ADMIRAL.

The ingredients used were:

	<u>Parts by</u> <u>Weight</u>
Polyol X	100
Water	2.6
Silicone surfactant	0.8
Triethylene diamine	0.33
Bis(2 dimethylamino)ether	0.15
Dibutyrtindilaurate	0.005
Vibration Control Agent (96% Butoxytriethylene glycol (4% Siloxane polyether block copolymer, (AB) <sub>n</sub> type and 150,000 to 200,000 M.W.	(See Table V)
Toluenediisocyanate	28.16
m-phenylenediisocyanate	7.04

Several different samples of foam were prepared with dimensions 40 x 40 x 10 cm. The moulded foams were crushed and the properties of the foams measured. The results are given in Table V.

Test D

This is a comparative test not according to the invention.

Foams were prepared as in Example 5 but without the used of vibration control agent.

The results obtained are shown in Table V.



TABLE V

Foam Characteristics												
Vibration control agents (parts by weight)	0	0	0	1	1	1	1	1	1	1	1	1
Overall density kg/m <sup>3</sup>	46.6	50.9	39.4	49.6	41.1	50.1	41.4	41.6	40.6			
Hardness (IFD, ASTM, 50%), N	300	356	189	296	203	232	222	213	217			
VC-Factor Qx	4.3	4.4	4.1	3.1	3.2	3.2	3.4	3.2	3.6			
Breathability Surface (Gurley), sec	1.0	2.0	1.0	2.0	2.0	3.0	3.0	3.0	2.0			
Breathability Core (NOPCO), SCFM	3.5	4.0	5.0	3.0	3.5	3.0	3.5	4.0	3.5			
Cell Size: mm	0.2-2.0	→	→	0.2-2.0	→	→	→	→	→			
No. of breaks in cell wall	8-12	→	→	8-12	→	→	→	→	→			
Internal residue cell surface %	40 to 60	→	→	60 to 70	→	→	→	→	→			

TABLE V - continuation

Foam Characteristics									
Vibration control agents (parts by weight)	1.5	1.5	1.5	2.0	2.0	2.0	2.5	2.5	2.5
Overall density kg/m <sup>3</sup>	50.8	42.7	51.2	50.2	40.4	41.1	50.7	42.2	50.0
Hardness, (IFD, ASTM, 50%), N	352	233	362	348	203	210	353	230	360
VC-Factor Qx	3.0	2.9	3.3	3.2	2.8	2.9	2.6	2.3	2.2
Breatability: Surface (Gurely), sec	4.0	3.0	4.0	5.0	4.0	4.0	5.0	4.0	5.0
Breatability: Core (NOPCO), SCFM	1.5	2.0	1.5	0.9	1.0	0.7	0.7	0.6	0.5
Cell Size: mm	0.2-2.5	→	→	0.2-2.5	→	→	0.2-3.5	→	→
No. of breaks in cell wall	6-12	→	→	6-12	→	→	6-10	→	→
Internal residue cell surface, %	65 to 75	→	→	70 to 85	→	→	80 to 90	→	→

In the Examples the samples used to determine foam properties were taken from the core of the foam as is usual for this type of measurement.

The values quoted for cell size and internal residual cell surface were determined by observations of a sample of individual cells

5 (usually about 20-25). The maximum and minimum cell size and residual cell surface were recorded.

The butoxytriethylene glycol used in the experiments in this specification was the mono ether.



## Claims:

1. A process for the production of a cold-cured moulded polyurethane flexible foam by reaction of a polyol with a polyisocyanate and foaming in the presence of a blowing agent followed by crushing of the moulded foam to open the foam cells characterised in that foaming takes place in the presence of a mixture of (1) a high molecular weight predominantly linear siloxane polyoxyalkylene block copolymer of formula  $(AB)_n$ , wherein  $n$  is an integer, A is a siloxane block and B is a polyoxyalkylene block and (2) a solvent which is compatible with the high molecular weight predominantly linear siloxane polyoxyalkylene block copolymer which has a viscosity of 3 to 1,000 cps, and a surface tension between 15 and 60 dynes/cm.
2. A process according to claim 1 for the production of a high resilience foam wherein at least 40% of the total polyol is constituted by a polyether triol having (a) an average primary hydroxyl content of at least 40 mole per cent, and (b) an average molecular weight of from 2,000 to 8,000.
3. A process according to either claim 1 or claim 2 characterised in that the high molecular weight predominantly linear siloxane polyoxyalkylene block copolymer is employed in admixture with lower molecular weight siloxane polyoxyalkylene condensation products derived from mono-ols.
4. A process according to any one of the preceding claims wherein the predominantly linear block copolymer is made by the reaction of a polyalkyl siloxane with a mixture of a polyalkylene ether diol and a mono-ol.
5. A process according to claim 4 wherein the mono-ol is an alkoxy ether derived from a polyalkylene ether diol.
6. A process according to either of claims 4 or 5 wherein the weight ratio of diol to mono-ol is in the range 1:1 to 1:2.

7. A process according to any one of the preceding claims wherein the polyoxyalkylene groups in the predominantly linear block copolymer are derived from polyoxyalkylene diols having average molecular weights in the range 1500 to 7000.
8. A process according to any one of the preceding claims wherein the average molecular weight of each siloxane block in the predominantly linear block copolymer is in the range 500 to 10,000.
9. The process according to any one of the preceding claims wherein the predominantly linear high molecular weight siloxane polyoxyalkylene block copolymer is a hydrolysable block copolymer.
10. The process according to any one of the preceding claims wherein the predominantly linear high molecular weight polyoxyalkylene block copolymers have an average molecular weight in the range 20,000 to 250,000.
11. The process according to claim 10 wherein the predominantly linear high molecular weight polyoxyalkylene block copolymer is a hydrolysable block copolymer and the average molecular weight is at least 65,000.
12. The process according to any one of the preceding claims wherein the solvent which is compatible with the high molecular weight predominantly linear siloxane polyoxyalkylene block copolymer has a viscosity of 10 to 500 cps.
13. A process according to any one of the preceding claims wherein the solvent has one to six reactive OH groups per molecule.
14. The process according to any one of the preceding claims wherein the solvent has a surface tension between 30 and 40 dynes/cm.
15. The process according to any one of the preceding claims wherein the solvent is a partially alkoxylated glycol.
16. The process according to claim 15 wherein the solvent is a butoxy mono ether of ethylene glycol.
17. The process according to any one of the preceding claims wherein the weight ratio of high molecular weight predominantly linear block copolymer to solvent is in the range 2:100 to 10:100.



# INTERNATIONAL SEARCH REPORT

International Application No PCT/GB 80/00072

<b>I. CLASSIFICATION OF SUBJECT MATTER</b> (if several classification symbols apply, indicate all) <sup>3</sup> According to International Patent Classification (IPC) or to both National Classification and IPC  Int. Cl. <sup>3</sup> : C 08 J 9/00; C 08 G 18/28																										
<b>II. FIELDS SEARCHED</b>  <div style="text-align: center;">Minimum Documentation Searched <sup>4</sup></div> <table style="width: 100%; border: none;"> <tr> <td style="width: 15%; border: none;">Classification System</td> <td style="border: none;">Classification Symbols</td> </tr> <tr> <td style="border: none;">Int. Cl.<sup>3</sup></td> <td style="border: none;">C 08 J 9/00; C 08 G 18/28</td> </tr> </table>			Classification System	Classification Symbols	Int. Cl. <sup>3</sup>	C 08 J 9/00; C 08 G 18/28																				
Classification System	Classification Symbols																									
Int. Cl. <sup>3</sup>	C 08 J 9/00; C 08 G 18/28																									
Documentation Searched other than Minimum Documentation to the Extent that such Documents are Included in the Fields Searched <sup>5</sup>  <div style="height: 40px;"></div>																										
<b>III. DOCUMENTS CONSIDERED TO BE RELEVANT</b> <sup>14</sup> <table border="1" style="width: 100%; border-collapse: collapse;"> <thead> <tr> <th style="width: 10%;">Category <sup>6</sup></th> <th style="width: 60%;">Citation of Document, <sup>14</sup> with indication, where appropriate, of the relevant passages <sup>17</sup></th> <th style="width: 30%;">Relevant to Claim No. <sup>18</sup></th> </tr> </thead> <tbody> <tr> <td></td> <td>US, A, 3793237, published February 19, 1974 see abstract, column 8, line 4 - column 12, line 17, see column 13, lines 8-42, see column 2, lines 60-68, J. Watkinson</td> <td>1, 2, 7-11</td> </tr> <tr> <td>A</td> <td>US, A, 3741917, published June 26, 1973, E.L. Morehouse</td> <td>1</td> </tr> <tr> <td></td> <td>US, A, 3836560, published September 17, 1974 see abstract, column 2, line 71 - column 4, line 75, column 12, lines 67-75, B. Prokai et al.</td> <td>1, 7-11</td> </tr> <tr> <td></td> <td>US, A, 4071483, published January 31, 1978 see abstract, column 17, line 65 - column 18, line 44, C.J. Litteral et al.</td> <td>1, 2, 12, 13, 15</td> </tr> <tr> <td></td> <td>US, A, 4031044, published June 21, 1977 see abstract and column 3, line 63 - column 4, line 20, W.G. Joslyn</td> <td>1, 2</td> </tr> <tr> <td>A</td> <td>FR, A, 2312520, published December 24, 1976 see claim 1, Naphtachimie</td> <td>12-16</td> </tr> <tr> <td></td> <td></td> <td>./.</td> </tr> </tbody> </table>			Category <sup>6</sup>	Citation of Document, <sup>14</sup> with indication, where appropriate, of the relevant passages <sup>17</sup>	Relevant to Claim No. <sup>18</sup>		US, A, 3793237, published February 19, 1974 see abstract, column 8, line 4 - column 12, line 17, see column 13, lines 8-42, see column 2, lines 60-68, J. Watkinson	1, 2, 7-11	A	US, A, 3741917, published June 26, 1973, E.L. Morehouse	1		US, A, 3836560, published September 17, 1974 see abstract, column 2, line 71 - column 4, line 75, column 12, lines 67-75, B. Prokai et al.	1, 7-11		US, A, 4071483, published January 31, 1978 see abstract, column 17, line 65 - column 18, line 44, C.J. Litteral et al.	1, 2, 12, 13, 15		US, A, 4031044, published June 21, 1977 see abstract and column 3, line 63 - column 4, line 20, W.G. Joslyn	1, 2	A	FR, A, 2312520, published December 24, 1976 see claim 1, Naphtachimie	12-16			./.
Category <sup>6</sup>	Citation of Document, <sup>14</sup> with indication, where appropriate, of the relevant passages <sup>17</sup>	Relevant to Claim No. <sup>18</sup>																								
	US, A, 3793237, published February 19, 1974 see abstract, column 8, line 4 - column 12, line 17, see column 13, lines 8-42, see column 2, lines 60-68, J. Watkinson	1, 2, 7-11																								
A	US, A, 3741917, published June 26, 1973, E.L. Morehouse	1																								
	US, A, 3836560, published September 17, 1974 see abstract, column 2, line 71 - column 4, line 75, column 12, lines 67-75, B. Prokai et al.	1, 7-11																								
	US, A, 4071483, published January 31, 1978 see abstract, column 17, line 65 - column 18, line 44, C.J. Litteral et al.	1, 2, 12, 13, 15																								
	US, A, 4031044, published June 21, 1977 see abstract and column 3, line 63 - column 4, line 20, W.G. Joslyn	1, 2																								
A	FR, A, 2312520, published December 24, 1976 see claim 1, Naphtachimie	12-16																								
		./.																								
<div style="font-size: small;"> <sup>9</sup> Special categories of cited documents: <sup>15</sup>  <div style="display: flex; justify-content: space-between;"> <div style="width: 45%;"> <sup>10</sup> "A" document defining the general state of the art  <sup>11</sup> "E" earlier document but published on or after the international filing date  <sup>12</sup> "L" document cited for special reason other than those referred to in the other categories  <sup>13</sup> "O" document referring to an oral disclosure, use, exhibition or other means           </div> <div style="width: 45%;"> <sup>16</sup> "P" document published prior to the international filing date but on or after the priority date claimed  <sup>17</sup> "T" later document published on or after the international filing date or priority date and not in conflict with the application, but cited to understand the principle or theory underlying the invention  <sup>18</sup> "X" document of particular relevance           </div> </div> </div>																										
<b>IV. CERTIFICATION</b> <table style="width: 100%; border: none;"> <tr> <td style="width: 40%; border: none;">           Date of the Actual Completion of the International Search <sup>1</sup>            June 26, 1980         </td> <td style="width: 60%; border: none;">           Date of Mailing of this International Search Report <sup>2</sup>            July 7, 1980         </td> </tr> <tr> <td style="border: none;">           International Searching Authority <sup>1</sup>            European Patent Office         </td> <td style="border: none;">           Signature of Authorized Officer <sup>20</sup>            G.L.M. KRUYDENBERG  </td> </tr> </table>			Date of the Actual Completion of the International Search <sup>1</sup> June 26, 1980	Date of Mailing of this International Search Report <sup>2</sup> July 7, 1980	International Searching Authority <sup>1</sup> European Patent Office	Signature of Authorized Officer <sup>20</sup> G.L.M. KRUYDENBERG																				
Date of the Actual Completion of the International Search <sup>1</sup> June 26, 1980	Date of Mailing of this International Search Report <sup>2</sup> July 7, 1980																									
International Searching Authority <sup>1</sup> European Patent Office	Signature of Authorized Officer <sup>20</sup> G.L.M. KRUYDENBERG																									

FURTHER INFORMATION CONTINUED FROM THE SECOND SHEET

- |   |       |
|---|-------|
| <p>A US, A, 3405077, published October 8, 1968<br/>see claims, A.J. Pastor et al.</p> | 12-16 |
| <p>A US, A, 3887483, published June 3, 1975,<br/>see abstract, E.L. Morehouse</p>     | 1     |

V. ☐ OBSERVATIONS WHERE CERTAIN CLAIMS WERE FOUND UNSEARCHABLE 10

This international search report has not been established in respect of certain claims under Article 17(2) (a) for the following reasons:

1. ☐ Claim numbers \_\_\_\_\_, because they relate to subject matter<sup>12</sup> not required to be searched by this Authority, namely:
  
2. ☐ Claim numbers \_\_\_\_\_, because they relate to parts of the international application that do not comply with the prescribed requirements to such an extent that no meaningful international search can be carried out<sup>13</sup>, specifically:

VI. ☐ OBSERVATIONS WHERE UNITY OF INVENTION IS LACKING 11

This International Searching Authority found multiple inventions in this international application as follows:

1. ☐ As all required additional search fees were timely paid by the applicant, this international search report covers all searchable claims of the international application.
2. ☐ As only some of the required additional search fees were timely paid by the applicant, this international search report covers only those claims of the international application for which fees were paid, specifically claims:
3. ☐ No required additional search fees were timely paid by the applicant. Consequently, this international search report is restricted to the invention first mentioned in the claims; it is covered by claim numbers:

Remark on Protest

- ☐ The additional search fees were accompanied by applicant's protest.
- ☐ No protest accompanied the payment of additional search fees.